6 mg 2 was heated with 0.5 ml Ac₂O at 80° for 1 hr. TLC gave 3.5 mg 4 and 1.5 mg 3 (¹H NMR spectra, see Table 1). 11α, 13-dihydrozaluzanin C (5). Colourless gum, IR $\nu_{\rm c}^{\rm CGL}$, cm⁻¹: 3470 (OH), 1785 (γ-lactone); MS m/z (rel. int.): 248 [M]⁺ (9), 230 [M - H₂O]⁺ (4), 177 (28), 157 (12), 152 (37), 134 (41), 121 (12), 105 (27), 81 (100).

$$[\alpha]_{24^{\circ}}^{\lambda} = \frac{589}{-12} \frac{578}{-10} \frac{546}{+4} \frac{436 \text{ nm}}{+86.4} (c = 0.25, \text{ CHCl}_3)$$

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GERMACRANOLIDES FROM DICOMA TOMENTOSA*

FERDINAND BOHLMANN, PAHUP SINGH and JASMIN JAKUPOVIC Institute for Organic Chemistry, Technical University of Berlin, D-1000 Berlin 12, West Germany

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Key Word Index—Dicoma tomentosa; Compositae; Mutisieae; sesquiterpene lactones; germacranolides; melampolides.

Abstract—An investigation of the aerial parts of *Dicoma tomentosa* afforded four new germacranolides and four melampolides together with urospermal A and its 11β ,13-dihydroderivative. The structures were elucidated by spectroscopic methods. The chemotaxonomic situation is discussed briefly.

INTRODUCTION

The genus *Dicoma* (Compositae, tribe Mutisieae) is placed in the subtribe Gochanatiinae [1] Chemical investigations have shown that 14, 15-oxygenated germacranolides are present [2] in one species, while two others only afforded acetylenic compounds and an allenic acid [2, 3]. We have now investigated a further species, *D. tomentosa*, which again gave germacranolides.

RESULTS AND DISCUSSION

While the roots of *Dicoma tomentosa* only afforded the triterpenes taraxasterol, stigmasterol, sitosterol and lupeyl acetate, the aerial parts gave

*Part 428 in the series "Naturally Occurring Terpene Derivatives". For Part 427 see Bohlmann, F., Singh, P., Joshi, K. C. and Singhi, C. L. (1982) Phytochemistry 21, 1441.

taraxasterol, urospermal A (1) [3] where the configuration of the 1,10 double bond has to be changed [Halsall T. G., unpublished], 11\(\beta\),13-dihydrourospermal A (10) [4] and eight further sesquiterpene lactones, the melampolides 2-5 and the germacranolides 6-9. The structure of 2 followed from the 'H NMR spectral data (Table 1) which were close to those of 1. As expected the H-15 signals were shifted downfield. The spectra of 8 and 9 (Table 1) showed that we were dealing with the mono- and diacetate of albicolide. All signals in the spectrum of 9 were assigned by spin decoupling. If the spectrum of 3 (Table 1) is compared with that of 8 the difference in the configuration of the 1, 10-double bond becomes obvious. The downfield shift of the H-1 signal in the spectrum of 3, if compared with the H-1 shift of 8, supported a cis-1, 10-double bond. The same is true in the spectrum of a diacetate isolated previously from a Dicoma species [2]. The configuration of the 1, 10double bond has to be changed to 1,10-cis. The spectra of 4 and 5 (Table 1) showed that these lacShort Reports 2123

Table 1. ¹H NMR spectral data of compounds 2-9 (400 MHz, CDCl₃, TMS as internal standard)

| | 2 | 3 | 4 | 5 | 6 | 7 | 8 | 9 |
|-------|------------------|------------|------------|-------------|------------|--|-----------|--|
| H-1 | 6.84 br dd | 5.56 br t | 6.53 ddd | 6.54 ddd | 6.24 br dd | 5.23 dd | 5.18 br t | 6.18 br t |
| H-2 | 2.61 m | | 2.52 m | 2.50 m | _ | 2.35 m | 2.33 m | 2.35 m |
| H-2' | 2.38 m | | 2.35 m | 2.42 m | _ | 2.27 m | _ | _ |
| H-3 | 2.67 ddd | 2.59 ddd | 2.64 ddd | 2.77 ddd | | 2.59 ddd | _ | 2.57 ddd |
| H-3' | 2.10 br dd | 1.85 dd | 2.10 br dd | 2.02 ddd | | 2.16 m | | 2.15 m |
| H-5 | 5.23 br d | 5.22 brd | 5.28 br d | 5.17 br d | 5.12 br d | 4.98 br d | 4.89 br d | 5.01 br d |
| H-6 | 4.60 dd | 4.83 dd | 4.67 dd | 4.78 dd | 4.51 dd | 4.73 dd | 4.80 dd | 4.62 dd |
| H-7 | 2.48 dddd | 2.60 ddddd | 2.28 dddda | 12.28 ddddd | 2.54 m | 2.93 dddd | 2.62 m | 2.63 ddddd |
| H-8) | 3.98 ddd | 2.30 m | 2.93 dddd | 2.92 dddd | _ } | 4.15 br dd | | 2.15 m |
| H-8′∫ | | 1.60 m | 1.53 m | 1.53 dddd | | | _ | 1.73 m |
| H-9) | 2.52 m | 2.20 m | 2.44 br dd | 2.41 ddd | <u> </u> | 2.87 brd | _ | 2.75 dd |
| H-9′∫ | | 2.05 m | 2.10 m | 2.22 ddd | | 2.46 dd | _ | 2.15 m |
| H-13 | 6.54 dd | 6.20 d | 6.19 d | 6.18 d | 6.29 d | 6.44 dd | 6.28 d | 6.31 d |
| H-13' | 6.33 dd | 5.45 d | 5.50 d | 5.50 d | 5.50 d | 6.37 dd | 5.53 d | 5.57 d |
| H-14) | 9.47 s | 4.62 d | 0.40 | 9.48 s | 9.90 br s | 4.57 dd | 4.55 dd | 4.50 d |
| H-14' | | 4.48 d | 9.48 s | | | 4.53 d | 4.39 d | 4.28 d |
| H-15 | 4.82 d 4.71 d | 4.49 d | 4.81 d | 4.48 d | 4.48 brs | 4.51 d | 4.21 br d | 4.55 d |
| H-15' | | 4.25 d | 4.77 d | 4.30 d | | 4.23 d | 4.09 br d | 4.60 d |
| OAc | 2.14 s | 2.05 : | 2.12 s | _ | 2.10 s | $\begin{cases} 2.14 \ s \\ 2.10 \ s \end{cases}$ | 2.06 s | $\begin{cases} 2.13 \ s \\ 2.08 \ s \end{cases}$ |
| ОН | 5.64 d | | | | | **, ** * | | |

J(Hz): 5, 6 = 6, 7 = 10; 14, 14' = 14; 15, 15' = 12.5; compound 2: 1, 2 = 7; 1, 2' = 9; 2, 3 = 6; 2', 3 = 2; 3, 3' = 12.5; 7, 8 = 10; 7, 13 = 3; 8, 9 = 3; 8, 9' = 4; 8, OH = 11; 13, 13' = 1.5; compound 3: 1, 2 = 8.5; 2, 3 = 6; 2', 3 = 2; 3' = 12; 3, 3' = 12.5; 7, 8 = 3; 7, 8' = 11; 7, 13 = 3.5; 7, 13' = 3; compound 4: 1, 2 = 7; 1, 2' = 9; 1, 14 = 1.5; 2, 3 = 6; 2', 3 = 2; 2', 3' = 3, 3' = 12; 7, 8 = 3; 7, 8' = 11; 7, 13 = 3.5; 7, 13' = 3; 8, 8' = 12; 8, 9 = 6.5; 8, 9' = 10; 9, 9' = 14; compound 5: 1, 2 = 7; 1, 2' = 9; 1, 14 = 1.5; 2, 3 = 6; 2, 3' = 2; 2', 3 = 2.5; 2', 3' = 3, 3' = 12; 7, 8 = 3; 7, 8' = 12; 7, 13 = 3.5; 7.13' = 3; 8, 9 = 6.5; 8, 9' = 12; 8, 8' = 14; 8', 9 = 2; 8', 9' = 6; 9, 9' = 14; 9', 14 = 1.5; compound 6: 1, 2 = 5; 1, 2' = 12.5; 7, 13 = 3.5; 7, 13' = 3; compound 7: 1, 2 = 5; 1, 2' = 2, 2' = 12; 2', 3 = 4; 2', 3' = 12; 2, 3 = 2.5; 3, 3' = 12; 7, 8 = 8, 9' = 10; 9, 9' = 13; 7, 13 = 3.5; 7, 13' = 3; 13, 13' = 1.5; compound 8 and 9: 1, 2 = 8.5; 2, 3 = 2', 3 = 3.5; 3, 3' = 12; 7, 8 = 4; 7, 8' = 10; 7, 13 = 3.5; 7, 13' = 3; 8, 9 = 6.5; 8, 9' = 10; 9, 9' = 13.

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tones only differed at C-15, the former being an acetate and the latter the corresponding hydroxy compound. The cis-configuration of the 1, 10-double bond followed from the chemical shift of H-14. All signals could be assigned by spin decoupling, although some signals were overlapping multiplets. Small differences in the conformation of 4 and 5 led to some unexpected differences, namely the presence of a clear allylic coupling in the spectrum of 5 not visible in that of 4. A slight downfield shift of H-3 in the spectrum of 5 was due to the more pronounced deshielding effect of the 15-hydroxy group. The ¹H NMR spectrum of 6 (Table 1) showed that this lactone was the 1, 10 trans-isomer of 4 as indicated by the chemical shift of H-14. The spectral data of 7 showed that this lactone was the 8α -hydroxy derivative of 9. As usual in 8α -hydroxy germacranolides the H-13' signal was shifted downfield and a coupling $J_{13,13}$ was visible. Also H-7 was slightly deshielded by the 8hydroxy group.

Further Dicoma species must be investigated to see whether sesquiterpene lactones of types 1-9 are characteristic for the genus and perhaps for the subtribe as similar lactones have been isolated from Actinoseris [4], Cnicothamnus [5], Gochnatia [6] and Wunderlichia [4], while from Pertya [7] guaianolides were reported. By contrast Oldenburgia and Onoseris [8] have so far yielded no lactones. However, the constituents of the latter species indicated that this genus may be better placed in the Nassauviinae, while those of Pleiotaxis showed that it perhaps should be placed in the Cynareae [9].

EXPERIMENTAL

The air-dried plant material (collected in Transvaal, voucher 81/42, deposited in the Botanic Research Institute, Pretoria) was extracted with Et₂O-petrol (1:2) and the resulting extract was separated by repeated TLC (Si gel). The aerial parts (10 g) afforded 20 mg taraxasterol, 5 mg 1, 10 mg 2, 1.5 mg 3, 3.5 mg 4, 3 mg 5, 1.5 mg 6, 4 mg 7, 1 mg 8, 6.4 mg 9 and 1.5 mg 10 (solvent: C₆H₆-CH₂Cl₂-Et₂O, 2:2:1), while the roots (2 g) gave 5 mg lupeyl acetate, 4 mg taraxasterol, 1 mg stigmasterol and 2 mg sitosterol. Known compounds were identified by comparing their ¹H NMR spectra with those of authentic material. Due to the small amounts no mp could be determined.

*Urospermal A-*15-O-acetate (2). Colourless solid, IR $\nu_{\rm max}^{\rm CHCl_3}$ cm⁻¹: 3420 (OH), 1780 (γ-lactone), 1750 (OAc), 1690, 1630 (C=CCHO); MS m/z (rel. int.): 320 [M]⁺ (0.5) (C₁₇H₂₀O₆), 260 [M – HOAc]⁺ (55), 242 [260 – H₂O]⁺ (25), 214 [242 – CO]⁺ (31), 213 [242 × CHO]⁺ (31), 69 (100).

$$[\alpha]_{24^{\circ}}^{\lambda} = \frac{589}{+91} \frac{578}{+97} \frac{546}{+111} \frac{436 \text{ nm}}{+218}$$
 CHCl₃; $c = 1.0$.

1, 10-cis-Albicolide-14-O-acetate (3). Colourless solid, IR $\nu_{\rm max}^{\rm CHCl_3}\,{\rm cm}^{-1}$: 3500 (OH), 1770 (γ -lactone), 1740 (OAc); MS m/z (rel. int.): 306 [M]+ (0.1), 246.126 [M – HOAc]+ (21) (C₁₅H₁₈O₃), 228 [246 – H₂O]+ (19), 91 (63), 81 (63), 53 (100); [α]_D = +8.5° (CHCl₃; c = 0.14).

8-Desoxy urospermal A-15-O-acetate (4). Colourless solid, IR $\nu_{\rm max}^{\rm CHCl_1}$ cm⁻¹: 1775 (γ -lactone), 1745 (OAc), 1690, 1630 (C=CCHO); MS m/z (rel. int.): 304 [M]⁺ (0.1), 244.110 [M - HOAc]⁺ (100), 216 [244 - CO]⁺ (24), 215 [244 - CHO]⁺ (37).

8-Desoxy urospermal A (5). Colourless solid, IR $\nu_{\rm max}^{\rm CHCl_3}\,{\rm cm^{-1}}$: 3500 (OH), 1770 (γ -lactone), 1690, 1630 (C=CCHO); MS m/z (rel. int.): 262 [M]⁺ (2), 244.110 [M – H₂O]⁺ (8) (C₁₅H₁₆O₃), 216 [244 – CO]⁺ (12), 215 [244 – CHO]⁺ (15), 53 (100).

$$[\alpha]_{24^{\circ}}^{\frac{1}{2}} = \frac{589}{-20} \frac{578}{-26} \frac{546}{-32} \frac{436 \text{ nm}}{-52} \text{ CHCl}_3; \ c = 0.26.$$

 8α -Hydroxy albicolide-14-O-acetate (7). Colourless solid, IR $\nu_{\rm max}^{\rm CHCls}$ cm $^{-1}$: 3500 (OH), 1760 (γ -lactone), 1730 (OAc); MS m/z (rel. int.): 305.139 [M – OAc] $^+$ (82) (C $_{17}$ H $_{21}$ O $_{5}$), 245 [305 – HOAc] $^+$ (16), 244 [M – 2 × HOAc] $^+$ (15), 227 [245 – H $_{2}$ O] $^+$ (15), 226 [244 – H $_{2}$ O] $^+$ (13), 55 (100).

$$[\alpha]_{24^{\circ}}^{\frac{1}{2}} = \frac{589}{+41} \frac{578}{+63} \frac{546}{+74} \frac{436 \text{ nm}}{+142} \text{ CHCl}_3; \ c = 0.32.$$

Albicolide-14-O-aceate (8). Colourless gum, IR $\nu_{\rm max}^{\rm CHCl_3}$ cm⁻¹: 3500 (OH), 1770 (γ-lactone), 1740 (OAc); MS m/z (rel. int.): 246.126 [M – HOAc]⁺ (4) (C₁₅H₁₈O₃), 228 [246 – H₂O]⁺ (4), 57 (100).

Albicolide diacetate (9). Colourless solid, IR $\nu_{\text{max}}^{\text{CCI}_k}$ cm⁻¹: 1780 (y-lactone), 1750 (OAc): MS m/z (rel. int.): 348 [M]⁺ (0.1), 289 [M – OAc]⁺ (100), 229 [289 – HOAc]⁺ (72), 228.115 [M – 2 × HOAc]⁺ (27) (C₁₅H₁₆O₂).

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